

Preparation of tertiary alcohols and γ -lactones from allylsilanes and anhydrides[†]

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(received 30 January 1995, accepted 15 May 1995)

Summary – The TiCl_4 -mediated reaction of allyltrimethylsilane and anhydrides yields alkyldiallylcarbinols. In the case of the diallylsilane 1,8-bis(trimethylsilyl)octa-2,6-diene and cyclic anhydrides, spiro-lactones resulting from a *gem*-diallylation are obtained with high stereoselectivity.

allylsilane / anhydride / dialkylation

Introduction

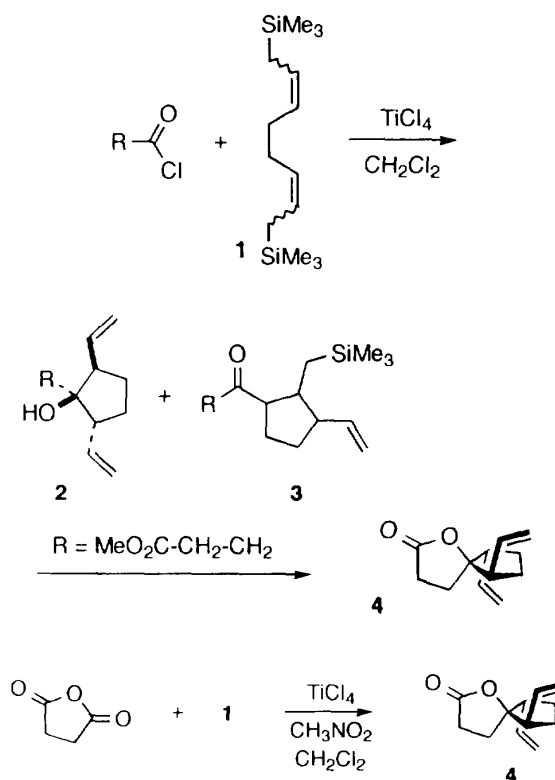
In 1995, Calas first showed that the electrophilic substitution of allylsilanes occurred with an allylic shift [1]. Twenty years later, Calas and Dunoguès demonstrated the regioselectivity of the acylation reaction by the report of a straightforward synthesis of artemisia ketone [2,3]. The *anti* stereospecificity of the reaction was later confirmed by Kumada [4]. The acylation reaction of allylsilanes is now part of the standard synthetic repertoire [5-9].

In 1988, we began the development of the chemistry of 1,8-bis(trimethylsilyl)octa-2,6-diene (BISTRO) **1** for the stereoselective preparation of potentially useful synthetic units [10]. We previously showed that the acylation of **1** with acyl chlorides led to a mixture of (*dl*)-1-alkyl-2,5-divinyl-cyclopentanol **2** and cyclopentylketones **3** or the spiro- γ -lactone **4** [11-13]. Compounds **2** and **4** are highly valuable in the synthesis of various substituted steroids or tetracyclic triterpenoids.

Acylation of allylsilanes with anhydrides has received little attention from fundamental and synthetic points of view [5,6,14-16]. We have shown that treatment of succinic anhydride with **1** in the presence of TiCl_4 and 4 equiv of nitromethane leads to **4** in a straightforward manner. The yield and the diastereoselectivity of the reaction are both excellent since the only *dl*-isomer is isolated in a 78% yield.

Encouraged by this result, we decided to condense allyltrimethylsilane and acetic anhydride. Tertiary alcohol **5** resulting from a twofold addition of an allyl group was isolated in a fair yield of 59% (alcohol **5** has previously been obtained in 90% yield by addition of allyl Grignard reagent to ethyl acetate [17]).

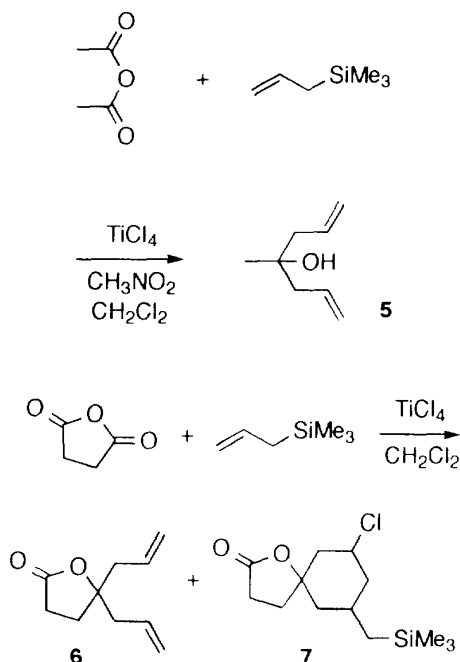
Using the same procedure, the bis allylic γ -lactone **6** was prepared by reaction of allyltrimethylsilane and



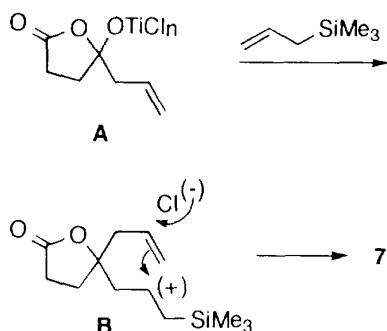
succinic anhydride (68% yield). When the reaction is carried out without nitromethane as a co-solvent, **6** is isolated in only 56% yield and a minor spirocyclic compound **7** (one isomer) is detected (11% yield).

[†] This article is dedicated to Professor Calas, in recognition of his outstanding research in organosilicon chemistry.

* Correspondence and reprints

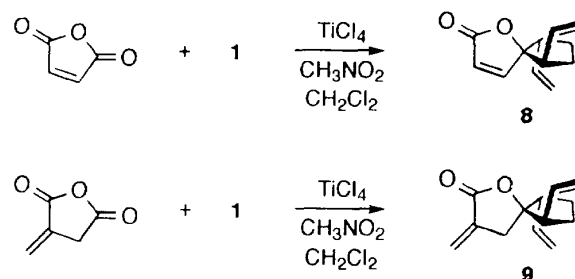


Compound **7** obviously arises from a participation reaction. We assume that the initial addition of allyltrimethylsilane to succinic anhydride affords the intermediate **A**. The second allyltrimethylsilane addition leads to the β -silyl carbocation **B**, which is set up to undergo a carbocationic cyclization process as depicted below :



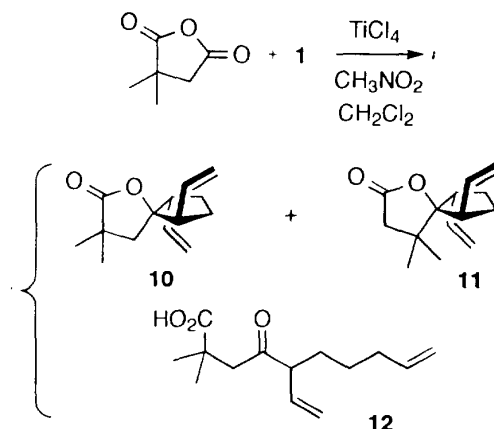
In order to extend the scope and limitations of the spirolactone formation, other anhydrides were tested in this reaction. High diastereoselectivity was also achieved with unsaturated cyclic compounds such as maleic anhydride or itaconic anhydride (methylenesuccinic anhydride). Unfortunately, the yields dramatically decrease to 26 and 13% yields for spirolactones **8** and **9** respectively (the other compounds are hydroxy and keto acids).

Dialkylation of 2,2-dimethylsuccinic anhydride leading to lactone **10** was obtained with high regioselectivity (**10/11** = 20:1) but in a 50% moderate yield. Acyclic ketoacid **12** was obtained as a byproduct from

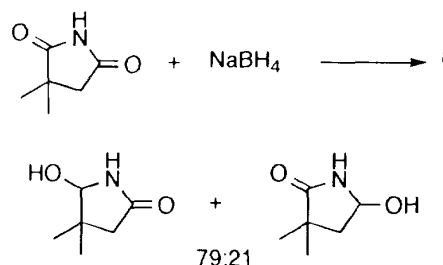


a monoalkylation process followed by a protolysis of the second allylsilane moiety (23% yield).

The stereochemistry of compounds **4** and **8-11** is deduced from presence of different signals for vinyl groups in ^{13}C and ^1H NMR spectra.

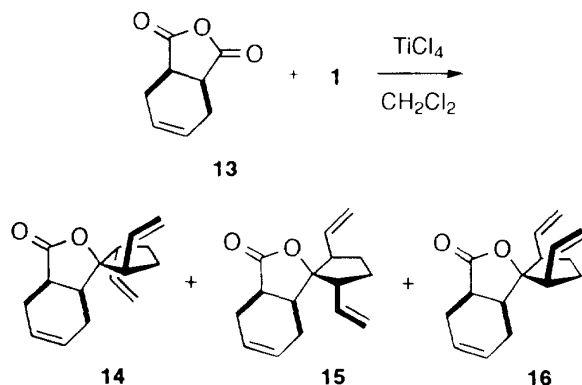


The high regioselectivity results from the addition to the less hindered carbonyl group. However, Speckamp has shown that the borohydride reduction of *gem*-disubstituted succinimides takes place preferentially at the seemingly more hindered carbonyl adjacent to the *gem* substituents. This remarkable regioselectivity was attributed to the reactant following a trajectory that is above, but not perpendicular to, the plane of the ring [18].

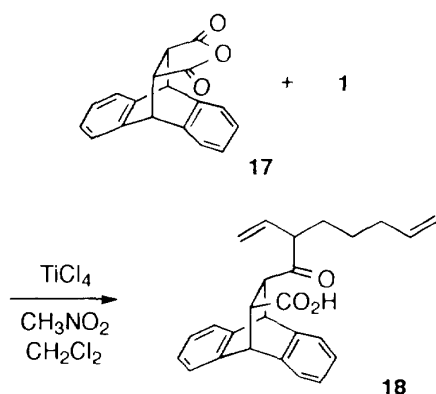


Comparison with our results suggests that addition of diallylsilane **1** to 2,2-dimethylsuccinic anhydride in the presence of TiCl_4 is dependent on several combined factors.

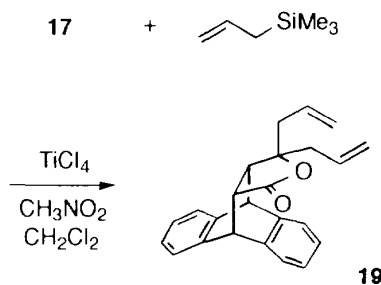
Substitution in the α - and β -position of succinic anhydride modifies the stereochemistry of the reaction. For instance, compound **13** led to the expected lactones **14** and **15** (inseparable mixture in 2:3 ratio, 35% yield) along with the unexpected lactone **16** (21% yield) in which the two vinyl groups are both on the same side of the spiro lactone as the oxygen atom.



The reaction of **1** with the Diels-Alder adduct **17** did not lead to the expected diallyllactone. Effectively, the presence of a very hindered face dramatically reduces the reactivity of this anhydride. Ketoacid **18**, which arises from a monoalkylation process followed by a protolysis, was the only compound isolated in 11% yield from the reaction mixture. In contrast, reaction of **17** with allyltrimethylsilane led to the dialkylated lactone **19** in a fair yield 54%.



Many desirable natural products contain γ -lactone moieties. These act as, for example, natural flavor components, pheromones [19] or lignans [20]. On the other hand, over the past decades, the problem of the five-membered ring construction has appeared in various guises due to its presence in numerous natural product series. Addition of BISTRO to cyclic anhydrides represents a solution for both of these challenges.



Experimental section

General

All reactions were run under argon in oven-dried glassware. TLC was performed on silica gel 60 F₂₅₄. ¹H and ¹³C NMR spectra were recorded in CDCl₃ solutions at 200 and 50 MHz respectively. Carbon-proton couplings were determined using DEPT sequence experiments [21]. Diastereoselectivity was determined by GC or ¹H NMR analyses prior to any purification. CH₂Cl₂ was distilled over P₂O₅. BISTRO was prepared according to a previously described procedure [10].

Representative procedure for the addition of allylsilanes to anhydrides

A three-necked flask equipped with a thermometer, septum cap, magnetic stirring bar, and argon outlet was charged with anhydrous CH₂Cl₂ (15 mL) and anhydrous nitromethane (2.2 mL, 40 mmol). The solution was cooled to -60°C. TiCl₄ was added (2.3 mL, 20 mmol) and then anhydride (10 mmol) in CH₂Cl₂ (2 mL). After 15 min of stirring at -70°C, the solution was cooled to -90°C and BISTRO (2.79 g, 11 mmol) or allyltrimethylsilane (2.39 mL, 21 mmol) in CH₂Cl₂ (3 mL) was added over 10 min. The resulting solution was warmed to -60°C and stirred for 15 h. The reaction was quenched by addition of aqueous saturated NH₄Cl solution (20 mL) and extracted with CH₂Cl₂ (3 × 15 mL). The extracts were washed until neutrality, dried over MgSO₄ and concentrated under vacuum. The residue was purified by chromatography on silica gel, eluted with a gradient of pentane/ether.

• (dl)-6,9-Divinyl-1-oxaspiro[4.4]nonan-2-one **4**

Using the procedure described above, reaction of 1.0 g of succinic anhydride and BISTRO gave 1.50 g (78% yield) of **4** after purification.

4 IR 1 774, 1 638, 1 234 cm⁻¹.

¹H NMR (CDCl₃) δ 5.85 (2H, m), 5.05 (4H, m), 2.75 (1H, dd, J = 7.6, 4.1 Hz), 2.48 (1H, m), 2.48 (2H, m), 2.22-1.66 (6H, m).

¹³C NMR (CDCl₃) δ 176.4 (s), 138.2 (d), 135.9 (d), 118.3 (t), 116.9 (t), 97.1 (s), 53.4 (d), 53.1 (d), 29.1 (t), 28.9 (t), 28.8 (t), 26.7 (t).

• 4-Methylhepta-1,6-dien-4-ol **5**

Using the procedure described above, reaction of 0.945 mL of acetic anhydride and allyltrimethylsilane gave 0.74 g (59% yield) of **5** after purification [17].

5 IR 3 409, 1 644, 1 157 cm⁻¹.

¹H NMR (CDCl₃) δ 5.93-5.72 (2H, m), 5.12-5.02 (4H, m), 2.19 (4H, d, J = 7.4 Hz), 1.13 (3H, s).

¹³C NMR (CDCl₃) δ 133.9 (d)(2C), 118.7 (t)(2C), 77.1 (s), 46.19 (t)(2C), 26.7 (q).

• **5,5-Diprop-2-enyldihydrofuran-2(3H)-one 6**

Using the procedure described above, reaction of 1.0 g of succinic anhydride and allyltrimethylsilane gave 1.08 g (68% yield) of **6** after purification.

6 IR 1 772, 1 191 cm^{-1} .

^1H NMR (CDCl_3) δ 5.66 (2H, ddt, $J = 18.4, 8.9, 7.3$ Hz), 5.08–4.99 (4H, m), 2.42 (2H, t, $J = 8.55$ Hz), 2.34–2.26 (4H, m), 1.95 (2H, t, $J = 8.55$ Hz).

^{13}C NMR (CDCl_3) δ 176.3 (s), 131.4 (d)(2C), 119.5 (t)(2C), 86.7 (s), 43.1 (t)(2C), 28.8 (t), 28.6 (t).

• **7-Chloro-9-[(trimethylsilyl)methyl]-1-oxaspiro[4.5]decan-2-one 7**

Using the procedure described above, but without nitromethane, reaction of 1.0 g of succinic anhydride and allyltrimethylsilane gave 0.93 g (56% yield) of **6** and 0.30 g (11% yield) of **7** after purification.

7 M p 63°C .

IR (film) 1 780, 1 549, 764 cm^{-1}

^1H NMR (CDCl_3) δ 4.40 (1H, m), 2.54 (2H, t, $J = 8.35$ Hz), 2.37 (1H, m), 2.25 (1H, m), 2.09 (1H, m), 1.97 (2H, t, $J = 8.35$ Hz), 1.92 (1H, m), 1.89 (1H, m), 1.43 (1H, m), 1.19 (1H, m), 0.49 (2H, m), 0.01 (9H, s).

^{13}C NMR (CDCl_3) δ 176.3 (s), 84.7 (s), 54.9 (d), 46.7 (t), 42.7 (t), 42.5 (t), 35.1 (t), 28.0 (t), 24.6 (d), 24.1 (t), –0.49 (q).

• **(dl)-6,9-Divinyl-1-oxaspiro[4.4]non-3-en-2-one 8**

Using the procedure described above, reaction of 0.98 g of maleic anhydride and BISTRO gave 0.254 g (26% yield) of **8** after purification.

8 IR (film) 1 756, 1 639, 921 cm^{-1}

^1H NMR (CDCl_3) δ 7.21 (1H, d, $J = 5.63$ Hz), 6.04 (1H, d, $J = 5.63$ Hz), 5.63 (2H, m), 5.05 (4H, m), 2.83 (1H, m), 2.75 (1H, m), 2.22–1.66 (4H, m).

^{13}C NMR (CDCl_3) δ 171.7 (s), 156.8 (d), 137.1 (d), 133.5 (d), 121.3 (d), 118.0 (t), 116.4 (t), 98.9 (s), 53.1 (d), 50.0 (d), 29.7 (t), 29.0 (t).

Anal calc for $\text{C}_{12}\text{H}_{14}\text{O}_2$: C, 75.76; H, 7.42. Found: C, 75.47; H, 7.56.

• **(dl)-3-Methylene-6,9-divinyl-1-oxaspiro[4.4]nonan-2-one 9**

Using the procedure described above, reaction of 1.12 g of itaconic anhydride and BISTRO gave 0.145 g (13% yield) of **9** after purification.

9 IR 1 762, 1 639, 1 267 cm^{-1} .

^1H NMR (CDCl_3) δ 6.13 (1H, t, $J = 2.75$ Hz), 5.69 (2H, m), 5.52 (1H, m), 5.09 (4H, m), 2.97 (1H, 1/2AB, t, $J = 17.5, 2.75$ Hz), 2.63 (1H, 1/2AB, t, $J = 17.5, 2.75$ Hz), 2.8–2.57 (2H, m), 2.22–1.66 (4H, m).

^{13}C NMR (CDCl_3) δ 171.7 (s), 138.1 (d), 135.3 (s), 135.1 (d), 121.4 (t), 118.0 (t), 116.4 (t), 93.7 (s), 53.3 (d), 53.0 (d), 32.5 (t), 28.4 (t), 28.3 (t).

Anal calc for $\text{C}_{13}\text{H}_{16}\text{O}_2$: C, 76.44; H, 7.90. Found: C, 76.58; H, 8.15.

• **(dl)-3,3-Dimethyl-6,9-divinyl-1-oxaspiro[4.4]nonan-2-one 10, (dl)-4,4-dimethyl-6,9-divinyl-1-oxaspiro[4.4]nonan-2-one 11 and 2,2-dimethyl-4-oxo-5-vinyldec-9-enoic acid 12**

Using the procedure described above, reaction of 1.28 g of 2,2-dimethylsuccinic anhydride and BISTRO gave 0.614 g (48% yield) of **10**, 0.025 g (2%) of **11** and 0.510 g (23% yield) of **12** after purification.

10 IR 1 766, 1 238 cm^{-1} .

^1H NMR (200 MHz, CDCl_3) δ 5.62–5.37 (2H, m), 5.04–4.87 (4H, m), 2.63 (1H, td, $J = 8.6, 2.5$ Hz), 2.21 (1H, q, $J = 9.0$ Hz), 2.07 (1H, 1/2 AB, $J = 13.6$ Hz), 1.76 (1H, 1/2 AB, $J = 13.6$ Hz), 1.90–1.25 (4H, m), 1.12 (3H, s), 1.07 (3H, s).

^{13}C NMR (CDCl_3) δ 181.0 (s), 138.4 (d), 135.5 (d), 118.0 (t), 116.3 (t), 92.9 (s), 53.2 (d), 52.2 (d), 39.5 (s), 38.5 (t), 28.3 (t), 27.7 (t), 26.0 (q), 25.4 (q).

Anal calc for $\text{C}_{14}\text{H}_{20}\text{O}_2$: C, 76.33; H, 9.15. Found: C, 76.28; H, 9.26.

11 IR 1 769, 1 212 cm^{-1} .

^1H NMR (200 MHz, CDCl_3) δ 5.81 (1H, dt, $J = 17.04, 9.82$ Hz), 5.76 (1H, dt, $J = 17.2, 9.57$ Hz), 5.22–4.98 (4H, m), 2.95 (1H, q, $J = 9.57$ Hz), 2.85 (1H, dd, $J = 9.4, 7.5$ Hz), 2.72 (1H, 1/2 AB d, $J = 17.1, 0.87$ Hz), 2.05 (1H, 1/2 AB, $J = 17.1$ Hz), 2.10–1.40 (4H, m), 1.19 (3H), 1.16 (3H).

^{13}C NMR (CDCl_3) δ 176.2 (s), 138.3 (d), 137.0 (d), 117.7 (t), 116.5 (t), 95.1 (s), 51.8 (d), 47.9 (d), 46.2 (t), 40.5 (s), 29.1 (t), 28.5 (t), 25.4 (q), 21.7 (q).

12 IR (neat) 3 500–2 500, 1 787, 1 226, 1 018, 913 cm^{-1} .

^1H NMR (200 MHz, CDCl_3) δ 5.82–5.48 (2H, m), 5.10–4.86 (4H, m), 2.97 (1H, m), 2.78–2.71 (2H, m), 1.96 (2H, m), 2.00–1.20 (4H, m), 1.17 (3H, s), 1.15 (3H, s).

^{13}C NMR (CDCl_3) δ 208.5 (s), 184.0 (s), 138.2 (d), 136.0 (d), 118.1 (t), 114.7 (t), 57.4 (d), 50.7 (t), 39.6 (s), 34.4 (t), 30.0 (t), 26.1 (t), 25.4 (q), 25.1 (q).

• **(2S*,5S*,3'aR*,7'aS*)-2,5-Divinyl-3'a,4',7',7'a-tetrahydrospiro[cyclopentane-1,1'-isobenzofuran]-3(1H)-one 14 and (2R*,5R*,3'aR*,7'aS*)-2,5-divinyl-3'a,4',7',7'a-tetrahydrospiro[cyclopentane-1,1'-isobenzofuran]-3(1H)-one 15 and (2S*,5R*,3'aR*,7'aS*)-2,5-dimethyl-3'a,4',7',7'a-tetrahydrospiro[cyclopentane-1,1'-isobenzofuran]-3(1H)-one 16**

Using the procedure described above, reaction of 1.52 g of *cis*-1,2,3,6-tetrahydrophthalic anhydride and BISTRO gave 0.53 g (35% yield) of **14** and **15**, and 0.319 g (21% yield) of **16** after purification.

14 and **15**. IR 1 769, 1 216, 1 173, 921, 674 cm^{-1} .

^1H NMR (200 MHz, CDCl_3) δ 5.89–5.52 (2H, m), 5.16–4.80 (4H, m), 2.81 (1H, dt, $J = 6.7, 3.0$ Hz), 2.64 (2H, m), 2.20 (1H, dt, $J = 7.3, 3.0$ Hz), 2.12 (4H, m), 2.04–1.50 (5H, m).

^{13}C NMR (CDCl_3) δ 178.9 and 178.3 (s), 137.8 and 137.4 (d), 137.2 and 136.0 (d), 127.1 and 125.8 (d), 125.3 and 124.8 (d), 117.6 and 116.5 (t), 115.6 and 115.4 (t), 95.9 and 95.8 (s), 56.0 and 54.2 (d), 49.3 and 48.2 (d), 39.6 and 38.5 (t), 37.6 and 37.1 (t), 30.8 and 30.7 (d), 28.8 and 28.5 (d), 22.75 and 22.7 (t), 22.2 and 21.7 (t).

Anal calc for $\text{C}_{16}\text{H}_{20}\text{O}_2$: C, 76.65; H, 8.25. Found: C, 76.57; H, 8.30.

16 IR 1 767, 1 265, 738 cm^{-1} .

^1H NMR (200 MHz, CDCl_3) δ 5.92–5.72 (2H, m), 5.65 (2H, t, $J = 2.5$ Hz), 5.21–5.02 (4H, m), 3.31 (1H, tm, $J = 9.0$ Hz), 2.87 (1H, t, $J = 7.7$ Hz), 2.74 (1H, q, $J = 9.3$ Hz), 2.49–2.13 (4H, m), 2.07–1.87 (2H, m), 1.75–1.23 (3H, m).

^{13}C NMR (CDCl_3) δ 164.3 (s), 129.3 (d), 128.8 (d), 118.2 (d), 118.1 (d), 111.6 (t), 111.4 (t), 95.3 (s), 55.5 (d), 54.6 (d), 44.3 (t), 43.2 (t), 36.4 (d), 34.9 (d), 30.6 (t), 30.0 (t).

• 12-(2-Vinylheptanoyl)-9,10-dihydro-9,10-ethanoanthracene-11-carboxylic acid **18**

Using the procedure described above, reaction of 2.76 g of adduct **17** and BISTRO gave 0.409 g (11% yield) of **18** after purification.

18 IR 3 420, 1 745, 1 641, 1 050 cm^{-1} .

^1H NMR (200 MHz, CDCl_3) δ 7.39-7.07 (8H, m), 5.90-5.55 (2H, m), 5.28-5.03 (4H, m), 4.67 (1H, d, $J = 4.10$ Hz), 4.48 (1H, d, $J = 2.25$ Hz), 3.37 (1H, dd, $J = 9.61$, 4.10 Hz), 2.78-2.52 (3H, m), 2.28 (2H, d, $J = 7.18$ Hz).

^{13}C NMR (CDCl_3) δ 175.4 (s), 144.1 (s), 141.8 (s), 141.4 (s), 139.7 (s), 132.5 (d), 131.0 (d), 126.63 (d), 126.56 (d), 126.4 (d), 126.3 (d), 125.9 (d), 125.4 (d), 124.4 (d), 123.3 (d), 121.0 (t), 119.3 (t), 87.3 (s), 49.1 (t), 48.8 (t), 45.4 (d), 45.1 (d), 38.1 (d).

• 14,14-Diallyl-9,10,14,15-tetrahydro-9,10[3',4'] furanoanthracene-12(11H)-one **19**

Using the procedure described above, reaction of 2.76 g of adduct **17** and allyltrimethylsilane gave 1.85 g (54% yield) of **19** after purification.

19 IR 1 755, 1 208, 929, 744 cm^{-1} .

^1H NMR (200 MHz, CDCl_3) δ 7.36-7.26 (4H, m), 7.19-7.07 (4H, m), 5.90-5.54 (2H, m), 5.28-5.03 (4H, m), 4.67 (1H, d, $J = 4.10$ Hz), 4.48 (1H, d, $J = 2.25$ Hz), 3.37 (1H, dd, $J = 9.63$, 4.10 Hz), 2.78-2.52 (2H, m), 2.64 (1H, dd, $J = 9.63$, 2.25 Hz), 2.28 (2H, d, $J = 7.18$ Hz).

^{13}C NMR (CDCl_3) δ 175.4 (s), 144.1 (s), 141.8 (s), 141.4 (s), 139.7 (s), 132.5 (d), 131.0 (d), 126.6 (d), 126.5 (d), 126.4 (d), 126.3 (d), 125.9 (d), 125.4 (d), 124.4 (d), 123.3 (d), 120.9 (t), 119.3 (t), 87.3 (q), 49.1 (t), 48.8 (t), 45.4 (d), 45.2 (d), 45.1 (d), 38.1 (d).

Anal calc for $\text{C}_{24}\text{H}_{22}\text{O}_2$: C, 84.18; H, 6.24. Found: C, 84.27; H, 6.18.

Acknowledgment

We thank R Faure for assistance with NMR spectral analyses.

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- A leading review (*cf.* ref 6) has reported only two examples of reaction between allylsilanes and anhydrides. Calas has shown that 1,1-dimethylsilylcyclopent-3-enes react with acetic anhydride in the presence of boron trifluoride etherate to give β -fluorosilylated ethylenic ketones (*cf.* ref 14). Magnus has obtained an acetylcycloheptene by treatment of a trimethylsilylcycloheptene with acetic anhydride/acetic acid containing perchloric acid (*cf.* ref 15).
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